Physics Time-Line 585BC to 2000

Philip Gibbs (compiled 1995-2000)

From the Greek philosophers to string theorists, this is the chronology of discoveries in physics and cosmology. According to Legend, Archimedes discovered the principle of buoyancy while taking a bath. He jumped out and ran through the streets shouting "Eureka!" The scientific revolution took off 1800 years later after Gutenberg introduced the printing press in Europe and Coperincus broke the old cosmology and put humans in their place away from the centre of the universe. Since then, thousands of scientists have experienced that Eureka moment when they realised that they have seen a fundamental truth not known before.

These pages contain, in chronological order, a selective list of those discoveries from natural philosophy. These are the findings which have helped us understand the laws of physics, the universe and our place in it. The time-line is not yet complete and many more scientists will have the privilege of contributing to its future. Today the internet is taking over from print as the primary means of communicating scientific discoveries and anyone who wants to can participate.

- -585: Thales of Miletus, prediction of an eclipse
- -580: Thales of Miletus, birth of scientific thought
- -580: Thales of Miletus, water as the basic element
- -580: Thales of Miletus, magnets and attraction to rubbed amber
- -560: Thales of Miletus, first cosmologies
- -550: Anaximenes, flat Earth
- -525: Pythagoras, understanding the world and mathematics
- -520: Anaximander, Earth surface is curved (cylinder)
- -515: Parmenides, paradoxes of change and motion
- -500: Pythagoreans, Earth is a sphere
- -480: Oenopides, finds angle of Earth's tilt to ecliptic
- -480: Protagoras, reality comes from the senses
- -480: Heraclitus, fire as primary substance
- -480: Heraclitus, change is the essence of being
- -475: Parmenides, Earth is a sphere
- -470: Anaxagoras, materials are made of "seeds" (atoms)
- -470: Anaxagoras, sun, moon and stars are made of same material as Earth
- -470: Anaxagoras, sun as a hot glowing rock
- -460: Eudoxus, Celestial spheres
- -460: Empedocles, Four elements: Earth, Air, Fire and Water
- -455: Philolaus, Earth Rotates
- -450: Zeno, paradoxes of discrete or continuous space and time
- -445: Leucippus, indivisble atoms
- -425: Democritus, Atomic theory
- -390: Plato, theory of knowledge
- -390: Plato, ether as a fifth element
- -385: Democritus, Milky Way is composed of many stars
- -370: Aristotle, Free falling bodies accelerate but heavier bodies fall faster
- -360: Heracleides, Venus and Mercury orbit the sun

-352: Chinese, recorded observation of a supernova

-350: Heracleides, Rotation of the Earth

-340: Aristotle, Earth is a sphere

-340: Aristotle, Space is continuous and always filled with matter

-335: Kiddinu, precession of equinoxes

-335: Strato, experiments with falling bodies and levers

-330: Aristotle, physics and metaphysics

-330: Aristotle, geocentric cosmology

-325: Pytheas, tides are caused by moon

-306: Epicurus, support for atomic theory

-295: Euclid, elements of mathematics

-265: Zou Yan, five elements: water, metal, wood, fire and earth

-260: Aristarchus of Samos, ratio of Earth-Sun distance to Earth-Moon distance from angle at half moon

-260: Aristarchus of Samos, distance and size of moon from Earth's shadow during lunar eclipse

-260: Aristarchus of Samos, heliocentric cosmology

-250: Chinese, free bodies move at constant velocity

-240: Archimedes, Principle of levers and compound pulley

-240: Archimedes, Archimedes' principle of hydrostatics

-235: Eratosthenes, Measurement of Earth's circumference

-190: Seleucus, further support for heliocentric theory

-170: Chinese, record of sun spots

-150: Hipparchus, precession of the equinoxes

-130: Hipparchus, size of moon from parallax of eclipse

83: Chinese, loadstone compass

100: Bhaskara, diameter of the Sun

100: Hero of Alexandria, expansion of air with heat

100: Hero of Alexandria, laws of light reflection

130: Ptolemy, geocentric cosmology of epicycles

180: Egypt, alchemy

550: Johannas Philoponus, impetus keeps a body moving

721: Abu Hayyan, preparation of chemicals such as nitric acid

890: Al-Razi, atomic of matter and space

890: Al-Razi, Andromeda galaxy

1000: Ali Al-hazen, reflection, refraction and lenses

1000: Ali Al-hazen, pinhole camera to demonstrate that light travels in straight lines to the eye

1054: China and Arabia Supernova of Crab Nebula recorded

1121: Al-khazini gravity acts towards centre of Earth

1155: Bhaskara first description of a perpetual motion machine

1225: Jordanus Nemorarius, mechanics of lever and composition of motion

1250: Albertus Magnus, isolation of arsenic

1260: Roger Bacon, empiricism

1267: Roger Bacon, magnifying lens

1269: Pierre de Maricourt, experiments with magnets and compass

1304: Theodoric of Freibourg, experiments to investigate rainbows

1320: William of Occam, Occam's Razor

1355: Jean Buridan, physics of impetus

1440: Nicolas Cusanus, Earth is in motion

1440: Nicolas Cusanus, infinite universe

1450: Johann Gutenberg, first printing press in Europe

1472: Johannes Regiomontanus, observation of Halley's Comet

1480: Leonardo de Vinci, description of parachute 1480: Leonardo de Vinci, compares reflection of light to reflection of sound waves 1490: Leonardo de Vinci, capillary action 1492: Leonardo de Vinci, foresees flying machines 1494: Leonardo de Vinci, foresees pendulum clock 1514: Nicolaus Copernicus, writes about heliocentric theory but does not yet publish 1515: Leonardo Da Vinci, progress in mechanics, aerodynamics and hydraulics 1537: Niccolo Tartaglia, trajectory of a bullet 1551: Girolamo Cardano, studies of falling bodies 1553: Giambattista Benedetti, proposed equality of fall rates 1543: Nicolaus Copernicus, heliocentric theory published 1546: Gerardus Mercator, Magnetic pole of Earth 1572: Tycho Brahe, witnesses a supernova and cites it as evidence that the heavens are not changeless 1574: Tycho Brahe, Observes that a comet is beyond the moon 1576: Tycho Brahe, constructs a planetary observatory 1576: Thomas Digges, illustration of an infinite universe surrounding a Copernican solar system 1577: Tycho Brahe, observes that a comet passes through the orbits of other planets 1581: Galileo Galilei, constancy of period of pendulum 1581: Robert Norman, dip of compass shows that Earth is a magnet 1584: Giordano Bruno, suggests that stars are suns with other Earth's in orbit 1585: Giovanni Benedetti, impetus theory is better than Aristotle's physics 1585: Simon Stevin, law of equilibrium 1586: Simon Stevin, pressure in column of liquid 1586: Simon Stevin, verification of equality of fall rates 1589: Galileo Galilei, showed that objects fall at the same rate independent of mass 1592: Galileo Galilei, suggests that physical laws of the heavens are the same as those on Earth 1592: Galileo Galilei, primitive thermometer 1593: Johannes Kepler, related planets to platonic solids 1596: David Fabricius, observes a variable star, (Mira Ceta) 1600: Galileo Galilei, study of sound and vibrating strings 1600: William Gilbert, static electricity and magnetism 1604: Johannes Kepler, mirrors, lenses and vision 1604: Galileo Galilei, distance for falling object increases as square of time 1608: Hans Lippershey, optical telescope 1609: Lippershey and Janssen, the compound microscope 1609: Johannes Kepler, 1st and 2nd laws of planetary motion 1609: Thomas Harriot, maps moon using a telescope 1609: Johannes Kepler, notion of energy 1609: Galileo Galilei, builds a telescope 1610: Galileo Galilei, observes the phases of Venus 1610: Galileo Galilei, observes moons of Jupiter 1610: Galileo Galilei, observes craters on the moon 1610: Galileo Galilei, observes stars in the Milky Way 1610: Galileo Galilei, observes structures around Saturn 1611: Fabricius, Galileo, Harriot, Scheiner, sunspots 1611: Marco de Dominis, explanation of rainbows 1611: Johannes Kepler, principles of the astronomical telescope 1612: Simon Marius, Andromeda galaxy 1612: Galileo Galilei, hydrostatics 1613: Galileo Galilei, principle of inertia

1615: S. de Caus, forces and work 1618: Francesco Grimaldi, interference and diffraction of light 1619: Johannes Kepler, 3rd law of planetary motion 1619: Johannes Kepler, explains why a comets tail points away from the Sun 1619: Rene Descartes, vision of rationalism 1620: Francis Bacon, the empirical scientific method 1620: Francis Bacon, heat is motion 1620: Jan Baptista van Helmont, introduces the word "gas" 1621: Willebrod Snell, the sine law of refraction 1624: Galileo Galilei, theory of tides 1626: Godfried Wendilin, verification of Kepler's laws for moons of Jupiter 1630: Cabaeus, attraction and repulsion of electric charges 1631: Pierre Gassendi, observes a transit of Mercury 1632: Galileo Galilei, Galilean relativity 1632: Galileo Galilei, Support for Copernicus' heliocentric theory 1632: John Ray, water thermometer 1636: G. Pers de Roberval, gravitational forces are mutual attraction 1636: Marin Mersenne, speed of sound 1637: Rene Descartes, inertia, mechanistic physics 1637: Rene Descartes, refraction, rainbow and clouds 1638: Galileo Galilei, motion and friction 1639: Jeremiah Horrocks, observes a transit of Venus 1640: Evangelista Torricelli, theory of hydrodynamics 1641: Ferdinand II, sealed thermometer 1642: Blaise Pascal, mechanical calculator 1644: Evangelista Torricelli, mercury barometer and artificial vacuum 1645: Ismael Boulliau, inverse square law for central force acting on planets 1648: Blaise Pascal, explains barometer as a result of atmospheric pressure 1650: Otto von Guericke, demonstration of the power of vacuum using two large hemispheres and 8 horses 1654: Ferdinand II, sealed thermometer 1656: Christiaan Huygens, rings and moons of Saturn 1657: Christiaan Huygens, pendulum clock 1657: Pierre Fermat, Fermat's principle in optics 1659: Christiaan Huygens, surface features on Mars 1660: Otto von Guericke, electrostatic machine 1660: Robert Boyle, sound will not travel in a vacuum 1661: Robert Boyle, corpuscular theory of matter 1661: Robert Boyle, chemical elements, acids and alkalis 1662: Robert Boyle, Boyle's law for ideal gases relating volume to pressure 1663: Blaise Pascal, isotropy of pressure 1663: James Gregory, describes a reflecting telescope 1663: Huygens, Wallace and Wren, laws of elastic collisions 1664: Robert Hooke, the great red spot of Jupiter 1664: Rene Descartes, published support for Copernican theory Isaac Newton 1665: Isaac Newton, studies the principles of mechanics and gravity, mass and force 1665: Giovanni Cassini, rotation periods of Jupiter, Mars and Venus

1665: Francesco Grimaldi, his wave theory of light is published

1665: Hooke, Huygens, colours of oil film explained by wave theory of light and interference

1665: Robert Hooke, studies with a microscope

1665: Robert Boyle, air is necessary for candles to burn

1666: Robert Boyle, fluid experiments

1666: Isaac Newton, studies spectrum of light

1666: Isaac Newton, begins work on laws of mechanics and gravitation

1667: Jean Picard, observes anomalies in star positions which are later explained as aberration

1668: John Wallis, conservation of momentum

1668: Isaac Newton, reflecting telescope

1669: Erasmus Bartholin, describes double refraction caused by polarisation effects of Iceland feldspar

1669: Hennig Brand, element phosphorus

1669: Gottfreid Leibniz, first concepts of action

1670: Robert Boyle, produces hydrogen by reacting metals with acid

1671: Giovanni Cassini, accurate measurement of distance to Mars and scale of solar system

1672: Jean Richer, the period of a pendulum varies with latitude

1672: Isaac Newton, variation of pendulum is due to equatorial bulge

1673: Ignace Pardies, wave explanation for refraction of light

1673: Christiaan Huygens, laws of centripetal force

1674: Robert Hooke, attempt to explain planetary motion as a balance of centrifugal force and gravitational attraction

1675: Giovanni Cassini, Saturn has separated rings which must be composed of small objects 1675: Isaac Newton, delivers his theory of light

1676: Olaus Roemer, measured the speed of light by observing Jupiter's moons

1676: Robert Hooke, law of elasticity and springs

1676: Edme Mariotte, pressure is inversely proportional to volume (Boyle's law) and height of atmosphere

1678: Robert Hooke, inverse square law of gravity

1678: Christiaan Huygens, writes about wave theory of light

1679: Christiaan Huygens, polarisation of light

1680: Isaac Newton, demonstrates that inverse square law implies eliptical orbits

1684: Isaac Newton, inverse square law and mass dependence of gravity

1684: Gottfreid Leibniz, differential calculus

1687: Isaac Newton, publishes laws of motion and gravitation

1687: Isaac Newton, publishes analysis of sound propagation

1688: P. Varignon, addition of forces

1690: Christiaan Huygens, principle of Huygens, secondary waves

1690: John Locke, knowledge comes only from experience and sensations

1692: Richard Bentley, why do stars not fall together under gravitation?

1702: Francis Hauksbee, rarefied air glows during electrical discharge

1704: Isaac Newton, publishes corpuscular theory of light and colour

1705: Edmund Halley, noticed that three previous comets are the same and predicts its return in 1758

1709: Gabriel Fahrenheit, alcohol thermometer

1710: George Berkeley, idealist philosophy against materialist

1714: Gottfreid Leibniz, energy conservation

1714: Gottfreid Leibniz, rejection of absolute space and time

1714: Gabriel Fahrenheit, mercury thermometer

1718: Edmund Halley, measures proper motion of stars

1720: Edmund Halley, early form of Olbers' paradox

1721: George Berkeley, space exists because of matter in it

1724: Gabriel Fahrenheit, supercooling of water

1727: Stephen Hales, makes oxygen

1728: James Bradley, speed of light and stellar aberration

1729: Stephen Gray, conduction of electricity

1731: Rene Reaumur, alcohol/water thermometer

1733: Charles Du Fay, recognises distinction between positive and negative electric charge

1735: Antonio de Ulloa, element platinum

1736: Leonhard Euler, differential equations in mechanics

1738: Daniel Bernoulli, kinetic theory of gas

1738: Daniel Bernoulli, hydrodynamics

1739: Georg Brandt, element cobalt

1740: Pierre Bouguer, gravitational anomalies

1742: Anders Celsius, reverse centigrade temperature scale

1743: Jean Christin, Celsius temperature scale

1743: Jean d'Alembert, energy in Newtonian mechanics

1744: Pierre de Maupertuis, principle of least action

1744: Jean d'Alembert, theory of fluid dynamics

1744: Leonhard Euler, Euler-Lagrange equations

1744: Mikhail Lomonosov, heat is a form of motion

1745: von Kleist, van Musschenbroek, Leyden jar for electric charge storage

1746: Andreas Marggraf, rediscovery of element zinc

1746: Leonhard Euler, wave theory of light refraction and dispersion

1747: d'Alembert, Euler, solution of equations for vibrating string

1748: Mikhail Lomonosov, conservation of mass and energy

1749: Thomas Melvill, early spectroscopy and yellow line of sodium in salt

1750: Benjamin Franklin, theory of electricity and lightning

1750: John Michell, magnetic induction

1750: John Michell, inverse square law for magnetic fields

1750: Thomas Wright, Milky Way could be due to slab like distribution of stars

1751: Benjamin Franklin, electricity can magnetise needles

1751: Frederik Cronstedt, element nickel

1752: Jean d'Alembert, viscosity

1754: Joseph Black, discovery of carbon dioxide showing that there are gases other than air 1755: Immanuel Kant, theory that the universe formed from a spinning nebula in an infinite hierarchy

1756: William Cullen, evaporation causes cooling

1756: Mikhail Lomonosov, supports wave theory of light

1761: Joseph Black, discovery and measurements of latent and specific heats

1761: John Harrison, portable chronometer

1765: Leonhard Euler, rigid body motions

1766: Joseph Priestley, inverse square law for electric charge

1766: Henry Cavendish, hydrogen is an element

1771: Luigi Galvani, electricity in animals

1772: Carl Scheele, saw air as two gases one of which encouraged combustion

1772: Daniel Rutherford, nitrogen

1772: Antoine Lavoisier, conservation of mass in chemical reactions

1772: Joseph Lagrange, theory of Lagrange points

1774: Priestley, Scheele, element oxygen

1774: Nevil Maskelyne, gravitational deflection of plumb line by a mountain

1774: Carl Scheele, element chlorine

1774: Johann Gahn, element manganese

1775: Alessandro Volta, electrical condenser

1776: Pierre-Simon Laplace, deterministic causality

1777: Antoine Lavoisier, composition of air and burning as a chemical reaction 1779: Charles Augustin de Coulomb, Coulomb's law of friction 1781: Immanuel Kant, Critique of pure reason 1781: William Herschel, discovery of Uranus 1781: Carl Scheele, element molybdenum in ore 1781: Charles Messier, catalogue of nebulae 1781: Heinrich Olbers, Uranus is a planet, not a comet 1782: Jacob Hjelm, isolation of element molybdenum 1782: Franz von Reichstein, element tellurium in ores 1782: William Herschel, catalog of double stars 1782: William Herschel, sun's motion through space 1783: John Michell, Newtonian black hole 1783: Fausto and Juan José de Elhuyar, element tungsten 1783: Rene Hauy, nature of crystals 1784: Henry Cavendish, water is a compound of oxygen and hydrogen 1784: Pierre Laplace, electrostatic potential 1785: Charles Augustin de Coulomb, electric force proportional to product of charges and inverse square of distance 1786: Antoine Lavoisier, distinction between elements and compounds 1787: Antoine Lavoisier, system for naming chemicals 1787: Jacques-Alexander Charles, law of gas expansion with temperature 1788: Joseph Lagrange, Lagrangian mechanics 1788: John Hunter, Diffusion of heat 1789: Antoine Lavoisier, Conservation of mass in chemical reactions 1789: Martin Klaproth, elements zirconium and uranium in compounds 1790: Definition of metric system in France 1790: Adair Crawford, element strontium in compounds 1791: William Gregor, element titanium in compounds 1794: Johann Gadolin, element yttrium in compounds 1794: Pierre Laplace, analysis of Newtonian black hole 1796: Alessandro Volta, chemical batteries and voltage 1797: Henry Cavendish, measured the gravitational constant with a torsion balance 1797: Nicholas Vauquelin, element beryllium identified in gem stones 1797: Nicholas Vauquelin, element chromium 1798: Benjamin Thompson, heat generated equals work done 1798: M. Klaproth, isolation of element tellurium 1798: Humphry Davy, Transmission of heat through vacuum 1798: Benjamin Rumford, experimental relation between work done and heat generated 1800: William Herschel, infrared rays from the Sun 1801: Johann Ritter, Ultraviolet rays 1801: Johann von Soldner, predicted Newtonian bending of light by sun 1801: Giuseppe Piazzi, first asteroid Ceres 1801: Humphry Davy, Electric arc 1801: Andres Manuel del Rio, compounds of element vanadium 1801: Charles Hatchett, element niobium in ores 1802: Heinrich Olbers, second asteroid Pallas 1802: Anders Ekeberg, element tantalum 1802: William Wollaston, dark lines in solar spectrum 1802: William Herschel, double stars are bodies in mutual orbit 1802: Thomas Young, interference and wave description of light 1802: Humphry Davy, Electrochemistry

1802: Joseph Gay-Lussac, Relation of Volume to Temperature of gases at fixed pressure 1803: William Wollaston, elements rhodium and palladium 1803: Smithson Tennant, elements osmium and iridium 1804: John Dalton, Law of partial pressures, Dalton's law 1807: Humphry Davy, isolation of elements sodium and potassium 1808: Humphry Davy, isolation of elements magnesium, strontium, barium and calcium 1808: Davy, Gay-Lussac and Thenard, isloation of element boron 1808: Joseph Gay-Lussac, Law of gas volumes in chemical reactions 1808: John Dalton, atomic theory of chemical reactions 1808: Etienne Malus, polarisation of reflected light 1809: Simeon-Denis Poisson, Poisson brackets in mechanics 1811: Amedeo Avogadro, molecular theory of gases and Avogadro's law 1811: Jean-Baptiste Fourier, harmonic analysis 1811: Bernard Courtois, element iodine 1812: David Brewster, behaviour of polarised light 1814: Joseph von Fraunhofer, spectroscope 1815: William Prout, atomic weights of elements are multiples of that for hydrogen 1815: Augustin Fresnel, theory of light diffraction 1816: Joseph von Fraunhofer, absorption lines in sun's spectrum 1817: Young and Fresnel, transverse nature of light 1817: Johan Arfvedson, element lithium 1817: Friedrich Strohmeyer, element cadmium 1817: Jöautns Berzelius, element selenium 1818: Augustin Fresnel, ether as absolute rest frame 1819: Dulong and Petit, relation of specific heats to atomic weight in 12 solid elements 1820: Andre Ampere, force on an electric current in a magnetic field 1820: Hans Christian Oersted, an electric current deflects a magnetised needle 1820: Biot and Savart, force law between an electric current and a magnetic field 1821: Thomas Seebeck, thermocouple and thermoelectricity 1821: Joseph von Fraunhofer, diffraction grating Michael Faraday 1821: Michael Faraday, plotted the magnetic field around a conductor 1821: Michael Faraday, first electric motor

1822: Andre Ampere, two wires with electric currents attract

1822: Charles Babbage, a prototype calculating machine

1822: Mary Mantell, first dinosaur fossil

1823: Michael Faraday, liquefies chlorine

1823: John William Herschel, suggests identification of chemical composition from spectrum

1823: William Sturgeon, electromagnets

1823: Heinrich Olbers, why is the sky dark?

1823: Johann Schweigger, galvanometer

1824: Sadi Carnot, Heat transfer goes from hot body to cold body

1824: Jöautns Berzelius, element silicon

1824: Jöautns Berzelius, isolation of element zirconium

1825: Hans Christian Oersted, isolation of element aluminium

1826: Antoine-J. Balard, element bromine

1827: Georg Ohm, electrical resistance and Ohm's law

1827: Robert Brown, Brownian motion

1828: Friedrich Wohler, isolation of element yttrium

1829: Johann Wolfgang, triads of chemical elements

1829: Thomas Graham, gas diffusion law

1829: Jons Berzelius, element thorium 1830: Charles Lyell, proposition that Earth is several million years old 1830: Nils Sefstrom, rediscovery and naming of vanadium 1831: Michael Faraday, a moving magnet induces an electric current 1831: Michael Faraday, magnetic lines of force 1831: Michael Faraday, the electric dynamo 1831: Michael Faraday, the electric transformer 1833: Michael Faraday, laws of electrolysis 1833: Joseph Henry, self-inductance 1834: Emile Clapeyron, entropy 1834: John Scott Russell, observed solitary waves in a canal 1834: William Hamilton, Principle of least action and Hamiltonian mechanics 1834: Heinrich Lenz, Law of electromagnetic forces 1835: Gustav-Gaspard Coriolis, Coriolis force 1838: Bessel, Henderson, Struve, first measurements of distance to a star by parallax 1839: Karl Mosander, Lanthanum 1840: Rive Marcet anomalous specific heat of diamond 1840: Joule and Helmholtz electricity is a form of energy 1840: Auguste Comte suggests that nature and composition of stars will never be known 1841: Eugene-Melchoir Peligot isolation of element uranium 1842: Christian Doppler theory of Doppler Effect for sound and light 1842: Justin von Mayer Conservation of heat and mechanical energy 1843: James Joule mechanical and electrical equivalent of heat 1843: Howard Aiken first mechanical programable calculator 1844: Kark Klaus element 44, ruthenium 1845: Michael Faraday, rotation of polarised light by magnetism 1845: Christopher Buys-Ballet, confirmation of Doppler effect for sound using trumpeters on a train 1846: Adams, Le Verrier, predicted position of Neptune 1846: Gustav Kirchhoff, Kirchoff's laws of electrical networks 1846: William Thomson (Kelvin), Incorrectly estimates Earth to be 100 million years old by heat 1846: Jahanne Galle, Neptune 1847: Hermann von Helmholtz, conservation of energy in Newtionian mechanics and gravity 1848: William Thomson (Kelvin), absolute temperature scale 1848: James Joule average velocity of gas molecules from kinetic theory 1849: Armand Fizeau first accurate measurement of the velocity of light in the laboratory using a toothed wheel 1850: Rudolf Clausius, generalised second law of thermodynamics 1850: Jean Foucault, light travels slower in water than in air 1850: Michael Faraday, experiments to find link between gravity and electromagnetism fail 1851: William Thomson (Lord Kelvin), dynamical theory of heat 1851: William Thomson (Lord Kelvin), absolute zero temperature 1851: Armand Fizeau, velocity of light in moving medium 1851: Franz Neumann, laws of electric-magnetic induction 1851: Jean Foucault, demonstrates rotation of Earth with a pendulum 1852: Jean Foucault, first gyroscope 1852: Joule, Thomson, an expanding gas cools 1853: Anders Angstrom, measured hydrogen spectral lines 1854: Hermann von Helmholtz, Heat death of the universe 1854: Bernhard Riemann, possibility of space curvature on small or large scales 1854: George Airy, Estimate of Earth mass from underground gravity 1855: William Parsons, spiral galaxies

1855: James Clerk Maxwell, mathematics of Faraday's lines of force

1857: James Clerk Maxwell, nature of Saturn's rings

1858: Wallace and Darwin, natural selection of species

1858: Balfour Stewart, conjecture equivalent to Kirchoff's law

1859: Hittorf and Plucker, cathode rays

1859: Bunsen and Kirchhoff, measurement of spectral line frequencies

1859: Urbain Le Verrier, anomolous perihelion shift of Mercury

1860: Gustav Kirchhoff, Kirchoff's Law and black body problem

1860: Maxwell and Waterston, equipartition theorem of statistical mechanics

1861: von Bunsen, Kirchhoff, elements caesium and rubidium found in spectra

1861: William Crookes, element thallium found by its spectra

1861: Johann Madler, Olbers's paradox would be resolved if the universe had a finite age

1862: Anders Angstrom, observed hydrogen in the sun

1863: William Huggins, stellar spectra indicate that stars are made of same elements as found on Earth

1863: Reich, Richter, element indium from its spectra

1864: John Newlands, chemical law of octaves

1864: James Clerk Maxwell, equations of electromagnetic wave propagation in the ether

1865: Rudolf Clausius, introduction of the term entropy

1867: James Clerk Maxwell, statistical physics and thermal equilibrium

1867: Henry Roscoe, isolation of element vanadium

1868: Pierre-Jules Janssen, lines of helium observed in the sun's spectrum

1868: Lockyer, Crookes, element helium recognised and named

1868: William Huggins, Doppler shifts of stellar spectra

1869: Dmitri Mendeleyev, periodic table of elements

1871: Dmitri Mendeleyev, prediction of new elements such as scandium, germanium, technetium, francium and gallium

1871: Ludwig Boltzmann, classical explanation of Dulong-Petit specific heats

1871: Tyndall and Rayleigh, light scattering and why the sky is blue.

1872: Ludwig Boltzmann, H-theorem

1873: James Clerk Maxwell, electromagnetic nature of light and prediction of radio waves

1873: Johannes van der Waals, intermolecular forces in fluids

1874: George Stoney, estimated the unit of charge and named it the electron

1875: Heinrich Weber, specific heat curves of solids

1875: James Clerk Maxwell, atoms must have a structure

1875: Paul-Emile Lecoq de Boisbaudran, element gallium

1877: Johann Loschmidt, questions validity of second law for time symmetric dynamics

1877: Ludwig Boltzmann, Boltzmann's probability equation for entropy

1877: Asaph Hall, two moons of Mars

1877: Cailletet and Pictet, liquid oxygen and nitrogen

1878: Josiah Willard Gibbs, thermodynamics of chemistry and phase changes

1879: Josef Stefan, empirical discovery of total radiation law, (Stefan's law)

1879: Lars Fredrik Nilson, element scandium

1879: Willaim Crookes, cathode rays may be negatively charged particles

1879: Albert Michelson, improved measurements of the speed of light

1880: Pierre and Jacques Curie, piezoelectricity

1881: Albert Michelson, light interferometer and absence of ether drift

1881: Josiah Willard Gibbs, vector algebra

1883: Ivan Puluy, prior discovery of X-rays

1883: Thomas Edison, thermionic emission

1883: George Fitzgerald, theory of radio transmission

1884: Ludwig Boltzmann, Derivation of Stefan's law for black bodies 1885: Johann Balmer, empirical formula for hydrogen spectral lines 1885: James Dewar, vacuum flask 1886: Henri Moissan, fluorine 1886: Clemens Winkler, element germanium 1887: Heinrich Hertz, transmission, reception and reflection of radio waves 1887: Michelson and Morley, absence of ether drift 1887: Michelson and Morley, fine structure of hydrogen spectrum 1887: Hertz, Hallwachs, photoelectric effect 1887: Woldemar Voigt, anticipated Lorentz transform to derive Doppler shift 1889: George Fitzgerald, length contraction 1889: Rolond von Eotvos, torsion balance to test equivalence of inertial and gravitational mass 1890: Johannes Rydberg, empirical formulae for spectral lines and Rydberg constant 1892: Hendrick Lorentz, theory that electricity is due to charged particles 1893: Ernst Mach, influence of all the mass in the universe determines what is natural motion 1893: Wilhelm Wien, derivation of black body displacement law 1893: Oliver Lodge, ether could not be carried along by matter 1894: Rayleigh and Ramsey, element argon 1894: Heinrich Hertz, radio waves travel at speed of light and can be refracted and polarised 1894: James Dewar, liquid oxygen 1894: Pierre Curie, why are there no magnetic monopoles? 1895: , isolation of helium from uranium ore 1895: Wilhelm Roentgen, X-rays 1895: Korteweg and de Vries, Explanation of solitary waves 1895: Jean-Baptiste Perrin, Cathode rays are negative particles 1895: Pierre Curie, loss of magnetism at high temperature, (Curie point) 1895: Hendrick Lorentz, first form of Lorentz transformation 1895: Hendrick Lorentz, Electromagnetic force on a charged particle 1896: Wilhelm Wien, conjectured exponential black body law 1896: Pieter Zeeman, spectral line splitting by magnetic field 1896: Antoine Henri Becquerel, natural radioactivity in uranium ore 1897: Ludwig Boltzmann, time reversal symmetry of electromagnetism 1897: Friedrich Paschen, verification of Wien's black body law at long wavelengths 1897: Kaufmann, J.J. Thomson, measurement of electron charge to mass ratio by deflection of cathode rays 1897: Weichert, J.J. Thomson, conjectured existence of light electron 1898: James Dewar, liquid hydrogen 1898: Guglielmo Marconi, Transmission of signals across the English Channel 1898: Pierre and Marie Curie, separation of radioactive elements, radium and polonium 1898: Ramsey and Travers, neon, krypton, xenon 1898: Joseph Larmor, complete form of Lorentz transformation 1898: Henri Poincare, guestions absolute time and simultaneity 1898: Ernest Rutherford, alpha and beta radiation 1899: Joseph John Thomson, measurement of the charge and mass of the electron 1899: Andre Debierne, element actinium 1899: Max Planck, universal scale of measurement from fundamental constants 1900: Lord Rayleigh, statistical derivation of short wavelength black body law 1900: Ernest Rutherford, first determination of a radioactive half-life 1900: Antoine Henri Becquerel, suggests that beta rays are electrons 1900: Lummer, Pringsheim, Rubens, Kurlbaum, failure of Wien's black body law at short wavelengths 1900: Max Planck, light quanta in black body radiation, Planck's black body law and Planck's constant

1900: Paul Villard, gamma rays

1900: Friedrich Dorn, element 86, radon

1900: Pyotr Lebedev, radiation pressure measured

1901: Max Planck, determination of Planck's constant, Boltzmann's constant, Avogadro's number and the charge on electron

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